Pharmacokinetics	

by the end of this lecture you should be able to	
 explain how drugs move through animals decide on an appropriate route of administration for a drug 	

pharmacokinetics

- What the animal does to the drug
 Movement of the drug through the body
- Movement of the drug through the bod









basic assumptions drugs must cross membranes to get to target actions are proportional to plasma 	
 not always true! 	



routes of administration	
 enteral oral (po = per os) sublingual rectal 	

routes of administration

parenteral

- intravenous (iv)
- intramuscular (im)
- nb muscle becomes meat in food animals!
- subcutaneous (sc or SQ)
 intraperitoneal (ip)





routes of administration			
• parenteral			
 intravenous (iv) intramuscular (im) 			
 nb muscle becomes meat in food animals! subcutaneous (sc or SQ) 			
– intraperitoneal (ip)			

routes of administration

- inhalation
- topical
- onto skin
- intramammary
- intrauterine
 onto cornea
- transdermal
- nasal
- epidural / intrathecal



absorption dissolution movement out of site of administration movement into blood vessels 	

dissolution most drugs must dissolve in water and oil ionisation important pH important 		

dissolution	
 main factors 	
– pills	
coatings	
disintegrants	
• vehicle	
– all	
solute	
- rumen contents	
 interstitial fluid 	



injection formulation	
complexes with soluble carriers – cyclodextrins	
 polyvinyl pyrrolidine (PVP) propylene glycol used to get lipid soluble drugs into aqueus solution 	

drug delivery devices]		
 "solution" in silicone rubber very slow release osmotic pumps 			
 predictable slow release mechanical pumps variable rates of delivery can be computer controlled ± feedback 			



barriers to absorption

- after iv administration
- none
- after oral administration
- gastric mucosa
- endothelium
- after im or sc administration

- endothelium



•	drug 😑	•
diffusion across ipid membranes	\$	diffusion through aqueous channel
	carrier mediated	

effects of pH	
 most drugs are either weak bases or weak acids issigned forms are not listed coluble 	
HAH+ +A-	
BH ⁺ <u>→</u> H ⁺ + B	





Henderson Hasselbach equation]	
for acids $pH = pK_a + log \frac{A}{AH}$		
for bases $pH = pK_a + log \frac{B}{BH^+}$		
ie, when $pH = pK_a$, the drug is 50% ionised		

effects of pH • when pH < pKa, more protonated drug exists (AH & BH+) • when pH > pKa, more unprotonated	
drug exists (A⁻ & B)	

effects of pH bases are ionised in acid solutions acids are ionised in alkaling solutions	





other factors influencing oral absorption

- blood flow
- reduced in shock
- surface area
- intestine > stomach
 contact time
- reduced in vomiting & diarrhoea
- food
- drugs may bind to food
- carrier mediated transport
- both ways



iv "absorption" • absorption is bypassed by iv injection • rate of injection = rate of absorption • if rate of absorption is critical to the patient, iv infusion can be used	









absorption]		
most drugs must be absorbed to act iv administration bypasses absorption			
orption depends on lipid solubility ionisation Is are often formulated to provide			
absorption			